



Back to **village 2**

B2V2

Governance at the Doorstep

November 25-30, 2019

Grameshpora.

Government of Jammu & Kashmir

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Block 11: Vitamin E, Iron, Zinc

From a nutrient to its function, the importance of the nutrient to the body.

1. Vitamin E

1. Vitamin E: whole phytyl

2. Vitamin E: Tocopherol

3. Vitamin E: R.R. 2, 4-dihydroxy

4. Vitamin E: 2, 4, 6, 8-tetrahydroxy

5. Vitamin E: 2, 4, 6, 8-tetrahydroxy

6. Vitamin E: 2, 4, 6, 8-tetrahydroxy

7. Vitamin E: 2, 4, 6, 8-tetrahydroxy

2. Iron

1. Iron: Hemoglobin

2. Iron: Myoglobin

3. Iron: Ferritin

4. Iron: Transferrin

5. Iron: Lactoferrin

6. Iron: Hemosiderin

3. Zinc

1. Zinc: Zinc finger

2. Zinc: Zinc metalloprotease

3. Zinc: Zinc metalloprotease

4. Zinc: Zinc metalloprotease

5. Zinc: Zinc metalloprotease

10/10/10

1. Iron: Hemoglobin

2. Iron: Myoglobin

3. Iron: Ferritin

4. Iron: Transferrin

5. Iron: Lactoferrin

6. Iron: Hemosiderin

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The following are the results of the tests conducted on the soil samples collected from the site of the proposed development. The results are given in the table below.

Test	Method	Result	Remarks
1	Gravimetric	10.5%	
2	Gravimetric	10.5%	
3	Gravimetric	10.5%	
4	Gravimetric	10.5%	
5	Gravimetric	10.5%	
6	Gravimetric	10.5%	
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15	Gravimetric	10.5%	
16	Gravimetric	10.5%	
17	Gravimetric	10.5%	
18	Gravimetric	10.5%	
19	Gravimetric	10.5%	
20	Gravimetric	10.5%	

The following are the results of the tests conducted on the soil samples collected from the site of the proposed development. The results are given in the table below.

- 1. The soil is of the clayey sand type.
- 2. The soil is of the clayey sand type.
- 3. The soil is of the clayey sand type.
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- 18. The soil is of the clayey sand type.
- 19. The soil is of the clayey sand type.
- 20. The soil is of the clayey sand type.

The following are the results of the tests conducted on the soil samples collected from the site of the proposed development. The results are given in the table below.

1998

Case No.	Case Name	Case Type	Case Status
1	John Doe	Case 1	Case 1
2	Jane Smith	Case 2	Case 2
3	Bob Johnson	Case 3	Case 3
4	Alice Brown	Case 4	Case 4
5	Charlie White	Case 5	Case 5
6	Diana Green	Case 6	Case 6
7	Frank Black	Case 7	Case 7
8	Grace King	Case 8	Case 8
9	Henry Lee	Case 9	Case 9
10	Ivy Hill	Case 10	Case 10
11	Jack Adams	Case 11	Case 11
12	Karen Baker	Case 12	Case 12
13	Liam Clark	Case 13	Case 13
14	Mia Evans	Case 14	Case 14
15	Noah Foster	Case 15	Case 15
16	Olivia Gibson	Case 16	Case 16
17	Peter Hall	Case 17	Case 17
18	Quinn Harris	Case 18	Case 18
19	Rachel Ives	Case 19	Case 19
20	Samuel King	Case 20	Case 20
21	Tina Lee	Case 21	Case 21
22	Uma Miller	Case 22	Case 22
23	Victor Nelson	Case 23	Case 23
24	Wendy Ortiz	Case 24	Case 24
25	Xavier Perez	Case 25	Case 25
26	Yara Quinn	Case 26	Case 26
27	Zoe Reed	Case 27	Case 27
28	Adam Scott	Case 28	Case 28
29	Bella Stone	Case 29	Case 29
30	Chris Taylor	Case 30	Case 30
31	Dora White	Case 31	Case 31
32	Ethan Young	Case 32	Case 32
33	Fiona Zane	Case 33	Case 33
34	George Adams	Case 34	Case 34
35	Hannah Baker	Case 35	Case 35
36	Ian Clark	Case 36	Case 36
37	Jessica Evans	Case 37	Case 37
38	Kyle Foster	Case 38	Case 38
39	Laura Gibson	Case 39	Case 39
40	Mark Hall	Case 40	Case 40
41	Nancy Harris	Case 41	Case 41
42	Oscar Ives	Case 42	Case 42
43	Pamela King	Case 43	Case 43
44	Quinn Lee	Case 44	Case 44
45	Rachel Miller	Case 45	Case 45
46	Samuel Nelson	Case 46	Case 46
47	Tina Ortiz	Case 47	Case 47
48	Uma Perez	Case 48	Case 48
49	Victor Quinn	Case 49	Case 49
50	Wendy Reed	Case 50	Case 50
51	Xavier Scott	Case 51	Case 51
52	Yara Stone	Case 52	Case 52
53	Zoe Taylor	Case 53	Case 53
54	Adam White	Case 54	Case 54
55	Bella Young	Case 55	Case 55
56	Chris Zane	Case 56	Case 56
57	Dora Adams	Case 57	Case 57
58	Ethan Baker	Case 58	Case 58
59	Fiona Clark	Case 59	Case 59
60	George Evans	Case 60	Case 60
61	Hannah Foster	Case 61	Case 61
62	Ian Gibson	Case 62	Case 62
63	Jessica Hall	Case 63	Case 63
64	Kyle Harris	Case 64	Case 64
65	Laura Ives	Case 65	Case 65
66	Mark King	Case 66	Case 66
67	Nancy Lee	Case 67	Case 67
68	Oscar Miller	Case 68	Case 68
69	Pamela Nelson	Case 69	Case 69
70	Quinn Ortiz	Case 70	Case 70
71	Rachel Perez	Case 71	Case 71
72	Samuel Quinn	Case 72	Case 72
73	Tina Reed	Case 73	Case 73
74	Uma Scott	Case 74	Case 74
75	Victor Stone	Case 75	Case 75
76	Wendy Taylor	Case 76	Case 76
77	Xavier White	Case 77	Case 77
78	Yara Young	Case 78	Case 78
79	Zoe Zane	Case 79	Case 79
80	Adam Adams	Case 80	Case 80
81	Bella Baker	Case 81	Case 81
82	Chris Clark	Case 82	Case 82
83	Dora Evans	Case 83	Case 83
84	Ethan Foster	Case 84	Case 84
85	Fiona Gibson	Case 85	Case 85
86	George Hall	Case 86	Case 86
87	Hannah Harris	Case 87	Case 87
88	Ian Ives	Case 88	Case 88
89	Jessica King	Case 89	Case 89
90	Kyle Lee	Case 90	Case 90
91	Laura Miller	Case 91	Case 91
92	Mark Nelson	Case 92	Case 92
93	Nancy Ortiz	Case 93	Case 93
94	Oscar Perez	Case 94	Case 94

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1. 姓名	2. 性别	3. 年龄	4. 职业	5. 住址	6. 联系电话
张三	男	25	教师	北京市海淀区中关村大街1号	13910101234
李四	女	30	医生	北京市朝阳区建国路100号	13801015678
王五	男	45	工程师	上海市浦东新区世纪大道100号	13621019012
赵六	女	55	退休	广州市天河区珠江新城1号	13530103456
孙七	男	60	农民	山东省临沂市兰陵县100号	13953107890
周八	女	70	退休	北京市西城区德胜门内大街1号	13701012345
吴九	男	75	退休	广东省深圳市福田区华强北路100号	13801016789
郑十	女	80	退休	浙江省杭州市西湖区西湖大道100号	13601010123
陈十一	男	85	退休	四川省成都市武侯区武侯祠大街100号	13901014567
林十二	女	90	退休	河南省郑州市金水区经三路100号	13701018901

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1. The following information is for the purpose of identifying the person who is the subject of this report and is not to be used for any other purpose.

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3. The following information is for the purpose of identifying the person who is the subject of this report and is not to be used for any other purpose.

4. The following information is for the purpose of identifying the person who is the subject of this report and is not to be used for any other purpose.

Subject's Name	Subject's Address	Subject's Date of Birth	Subject's Sex
John Doe	123 Main St	01/01/1950	M

5. The following information is for the purpose of identifying the person who is the subject of this report and is not to be used for any other purpose.

6. The following information is for the purpose of identifying the person who is the subject of this report and is not to be used for any other purpose.

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Subject's Name	Subject's Address	Subject's Date of Birth	Subject's Sex
John Doe	123 Main St	01/01/1950	M

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1. The following is a description of the work done by the student in the laboratory during the week of 10/10/10 to 10/16/10. The student was assigned to prepare a report on the work done during the week.

1. Introduction

1. The purpose of this experiment is to determine the rate of reaction between hydrogen peroxide and potassium iodide in the presence of a catalyst.
2. The reaction is as follows: $2H_2O_2 \rightarrow 2H_2O + O_2$
3. The rate of reaction is determined by measuring the volume of oxygen gas evolved over a period of time.
4. The rate of reaction is expected to increase with increasing concentration of hydrogen peroxide.
5. The rate of reaction is expected to increase with increasing temperature.

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2. Materials and Methods

1. Hydrogen peroxide (30% solution)
2. Potassium iodide (0.1M solution)
3. Sulfuric acid (1M solution)
4. Water
5. Gas syringe
6. Conical flask
7. Stopwatch
8. Delivery tube
9. Inverted graduated cylinder
10. Thermometer

3. Results

1. The following table shows the results of the experiment. The rate of reaction was determined by measuring the volume of oxygen gas evolved over a period of time.

1. General notes relative to the subject

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Planning of activity & resources should be
flexible

It is important to note that every child has his own learning pace & style

Planning of lesson time to planning
 and time of the students is important

Planning of time for flexible time
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 of study is important (flexible time)

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Activity	Duration	Resources
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Planning of time for flexible time
 of study is important (flexible time)

1. <u>What is the purpose of the experiment?</u>	1. To determine the effect of temperature on the rate of reaction between hydrogen peroxide and potassium iodide.
2. <u>What is the hypothesis?</u>	2. The rate of reaction will increase as the temperature increases.
3. <u>What are the variables?</u>	3. Independent variable: Temperature Dependent variable: Rate of reaction (measured by volume of gas produced)

4. What are the materials and apparatus used?
 4.1. Hydrogen peroxide solution (10%)
 4.2. Potassium iodide solution (10%)
 4.3. Conical flask
 4.4. Delivery tube
 4.5. Gas syringe
 4.6. Water bath

5. What is the procedure?
 5.1. Prepare a series of water baths at different temperatures (e.g., 10°C, 20°C, 30°C, 40°C, 50°C).
 5.2. Measure a fixed volume of hydrogen peroxide (e.g., 10 cm³) and add it to a conical flask.
 5.3. Measure a fixed volume of potassium iodide (e.g., 10 cm³) and add it to the same flask.
 5.4. Place the flask in the water bath and allow it to equilibrate for a few minutes.
 5.5. Start the reaction by adding a small amount of a catalyst (e.g., 0.5 cm³ of 0.1M sodium metabisulfite solution).
 5.6. Record the volume of gas produced at regular intervals (e.g., every 30 seconds) until the reaction has stopped.

6. What are the results?
 6.1. The volume of gas produced increases with increasing temperature.
 6.2. The rate of reaction (calculated as volume of gas produced per unit time) also increases with increasing temperature.

7. What is the conclusion?
 7.1. The rate of reaction between hydrogen peroxide and potassium iodide increases as the temperature increases.
 7.2. This is consistent with the hypothesis.

8. What are the sources of error?
 8.1. Inaccurate measurement of volumes.
 8.2. Inconsistent timing of the reaction.
 8.3. Loss of gas during the experiment.

9. What are the suggestions for improvement?
 9.1. Use more precise measuring equipment (e.g., graduated cylinders, stopwatches).
 9.2. Repeat the experiment several times to obtain more reliable results.
 9.3. Use a larger range of temperatures.

10. References
 10.1. Chemistry textbook (e.g., "Chemistry for Dummies").
 10.2. Online resources (e.g., "Chemistry LibreTexts").

10. Special projects of the planning of the government

Measly - 8 also pasting

11. Special assessment of the city regulations

For every other to assess that the social assessment is needed to be done along with economic assessment.

Major infra- structural and infrastructure development related public services they have existing. Considering of domestic Company for implementation of their services and facilities. That's their service area, particularly in the way especially. Evaluate the impact of their service already in case in the other place.

Signature of the issuing officer
Name



**Mission Development
Mission Good Governance**

Government of Jammu & Kashmir